

A tutorial to set up alchemical free energy perturbation calculations in NAMD

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1. Introduction

The main goal of this tutorial is to provide a guidance when setting up free energy calculations of alchemical transformations within NAMD.¹ As has been commented on amply, such *in silico* experiments have not reached yet the maturity to be viewed as black-box, routine jobs.² Either the set-up, the sampling protocol, or the analysis of the result should be considered with great care. The paradigm chosen in NAMD for performing alchemical transformations is the so-called *dual-topology* approach,³ wherein both the initial state, *viz.* $\lambda = 0$, and the final state, *viz.* $\lambda = 1$, are defined concurrently. As the molecular dynamics (MD) simulation progresses, the potential energy function characteristic of $\lambda = 0$ is scaled into that representative of $\lambda = 1$. Whereas the initial and the final states do not see each other in the course of the transformation, they, however, interact with the environment. The implication of these conditions is that a list of excluded atoms should be defined in the `psf` topology file. At the present time, PSFGEN does not permit the construction of such a list in an easy and straightforward fashion. The set-up of the two examples described hereafter will, therefore, be done within CHARMM.^{4,5}

2. Setting up the system

Perhaps the simplest alchemical transformation one could imagine, the result of which is completely independent of the potential energy function utilized, is the *zero-sum* ethane \rightarrow ethane mutation,^{6,3} wherein a methyl group vanishes at one end of the molecule, while another one appears at the other end. The accuracy of the computed free energy depends solely upon the sampling protocol adopted, regardless of the force field employed.

As can be seen in Figure 1, the hybrid defined for this transformation is a propane molecule, consisting of a juxtaposition of two ethane fragments, with a common $-\text{CH}_2-$ moiety. To prevent the initial state, *viz.* $\lambda = 0$, from interacting with the final state, *viz.* $\lambda = 1$, the atoms pertaining to the latter should be excluded. In CHARMM, this can be achieved by declaring explicitly, next to the definition of each atom, those atoms of the hybrid molecule that should not interact. In the CHARMM definition below, atoms CF, HF1, HF2, HF3 and HF of the final state will not see

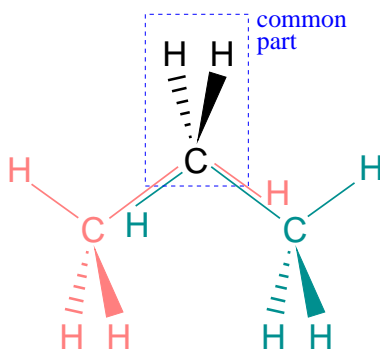


Figure 1: Dual-topology hybrid molecule used for the *zero-sum* ethane \rightarrow ethane alchemical transformation. The initial state, *viz.* $\lambda = 0$ (pink), and the final state *viz.* $\lambda = 1$ (cyan), are defined concurrently. The central $\text{--CH}_2\text{--}$ moiety is common to the two topologies.

atoms CI, HI1, HI2, HI3 and HI of the initial state. Alternatively, by uncommenting the list of atoms representative of the final state, and commenting out those of the initial state, the reverse transformation would take place.

```
read rtf card append
```

```
RESI ZERO          0.00          ! ethane -> ethane
GROUP
ATOM CI    CT3     -0.27         CF  HF1 HF2 HF3 HF  !
ATOM HI1   HA       0.09         CF  HF1 HF2 HF3 HF  !
ATOM HI2   HA       0.09         CF  HF1 HF2 HF3 HF  !
ATOM HI3   HA       0.09         CF  HF1 HF2 HF3 HF  !
GROUP
ATOM CM    CT3     -0.27          !      HI1      HM1    HF2    HF3
ATOM HM1   HA       0.09          !      \      |      |    /
ATOM HM2   HA       0.09          !      \HF     |      |    /
ATOM HI     HA       0.09         CF  HF1 HF2 HF3 HF  !      CI-----CM-----CF
ATOM HF     HA       0.09         !  CI  HI1 HI2 HI3 HI  !      /  |      |      HI\
GROUP
ATOM CF    CT3     -0.27         !  CI  HI1 HI2 HI3 HI  !      /  |      |      \
ATOM HF1   HA       0.09         !  CI  HI1 HI2 HI3 HI  !      HI2 HI3 HM2      HF1
ATOM HF2   HA       0.09         !  CI  HI1 HI2 HI3 HI  !
ATOM HF3   HA       0.09         !  CI  HI1 HI2 HI3 HI  !
BOND      CI  HI1         CI  HI2         CI  HI3          ! ethane 1
BOND      CF  HF1         CF  HF2         CF  HF3          ! ethane 2
BOND      CI  CM          CF  CM          ! common
BOND      CM  HM1         CM  HM2          ! common
BOND      CM  HI          ! ethane 1
```

```

BOND      CM  HF                                     ! ethane 2
ANGLE     HI1 CI  CM          HI2 CI  CM          HI3 CI  CM          ! ethane 1
ANGLE     HI1 CI  HI2        HI1 CI  HI3        HI2 CI  HI3          ! ethane 1
ANGLE     HF1 CF  CM          HF2 CF  CM          HF3 CF  CM          ! ethane 2
ANGLE     HF1 CF  HF2        HF1 CF  HF3        HF2 CF  HF3          ! ethane 2
ANGLE     CF  CM  HF          CI  CM  HI           ! common
ANGLE     HM1 CM  HM2                                     ! common
ANGLE     HM1 CM  CI          HM1 CM  CF           ! common
ANGLE     HM1 CM  HF          HM1 CM  HI           ! common
ANGLE     HM2 CM  CI          HM2 CM  CF           ! common
ANGLE     HM2 CM  HF          HM2 CM  HI           ! common
DIHEDRAL  HI1 CI  CM  HM1    HI1 CI  CM  HM2    HI1 CI  CM  HI      ! ethane 1
DIHEDRAL  HI2 CI  CM  HM1    HI2 CI  CM  HM2    HI2 CI  CM  HI      ! ethane 1
DIHEDRAL  HI3 CI  CM  HM1    HI3 CI  CM  HM2    HI3 CI  CM  HI      ! ethane 1
DIHEDRAL  HF1 CF  CM  HM1    HF1 CF  CM  HM2    HF1 CF  CM  HF      ! ethane 2
DIHEDRAL  HF2 CF  CM  HM1    HF2 CF  CM  HM2    HF2 CF  CM  HF      ! ethane 2
DIHEDRAL  HF3 CF  CM  HM1    HF3 CF  CM  HM2    HF3 CF  CM  HF      ! ethane 2

END

```

The second, less trivial application of alchemical free energy calculations consists in mutating in a short peptide the side chain of an amino acid. The example of the terminally blocked Ala–Ser–Ala tripeptide was chosen, in which the L–serine (Ser) residue was transformed into L–alanine (Ala).

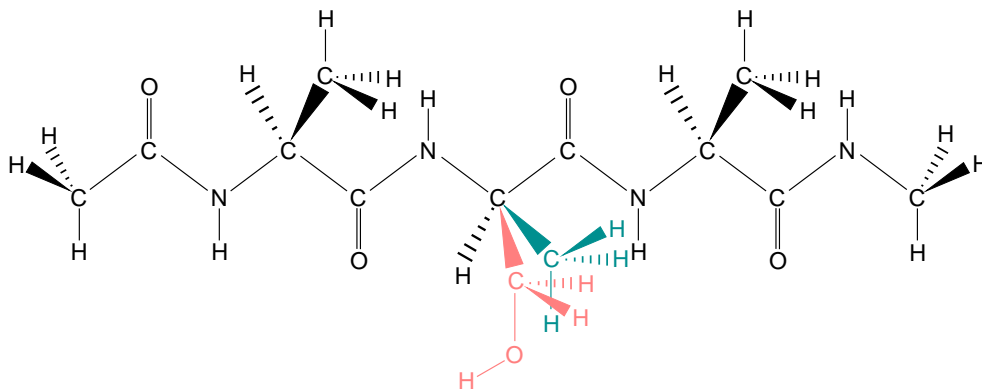


Figure 2: Dual-topology hybrid molecule used for the Ala–Ser–Ala \rightarrow (Ala)₃ alchemical transformation. The initial state, *viz.* $\lambda = 0$ (pink), and the final state *viz.* $\lambda = 1$ (cyan), are defined concurrently. Apart from the two side chains, the chemical groups of the tripeptide are common to the two topologies.

Here again, one must check that the side chain of the wild type does not see that of the mutant. To

achieve this condition, a hybrid amino acid should be defined, wherein those atoms characteristic of the final state, *viz.* $\lambda = 1$, are excluded from the list of atoms representative of the initial state, *viz.* $\lambda = 0$ — see Figure 2. In the CHARMM definition of the topology, atoms CBB, HB1B, HB2B and HB3B pertaining to Ala will not interact with atoms CBA, HB1A, HB2A, OGA and HG1A of Ser.

```

read rtf card append
! serine/alanine hybrid
27    1

DECL -CA
DECL -C
DECL -O
DECL +N
DECL +HN
DECL +CA

DEFA FIRS NTER LAST CTER
AUTO ANGLES DIHE

RESI STA          0.00
GROUP
ATOM N    NH1      -0.47      !      |
ATOM HN    H        0.31      !  N--N      HB1B
ATOM CA    CT1      0.07      !      |      HB1A
ATOM HA    HB        0.09      !      |      |      HB3
GROUP      ! HA--CA---CBA---OGA
ATOM CBA    CT2      0.05      CBB  HB1B HB2B HB3B      !      |      CBB      \
ATOM HB1A   HA        0.09      CBB  HB1B HB2B HB3B      !      |      |      HG1A
ATOM HB2A   HA        0.09      CBB  HB1B HB2B HB3B      !  O==C      HB2A
ATOM OGA    OH1     -0.66      CBB  HB1B HB2B HB3B      !      |
ATOM HG1A   H         0.43      CBB  HB1B HB2B HB3B      !  serine part
GROUP      !
ATOM CBB    CT3     -0.27      ! CBA  HB1A HB2A OGA  HG1A !
ATOM HB1B   HA        0.09      ! CBA  HB1A HB2A OGA  HG1A !
ATOM HB2B   HA        0.09      ! CBA  HB1A HB2A OGA  HG1A !
ATOM HB3B   HA        0.09      ! CBA  HB1A HB2A OGA  HG1A !  alanine part
GROUP
ATOM C      C         0.51
ATOM O      O        -0.51
BOND N      HN        N      CA      ! common
BOND C      CA        C      +N      CA      HA      ! common

```

```

BOND CBA  CA      OGA  CBA      OGA  HG1A      ! serine
BOND HB1A CBA      HB2A CBA                      ! serine
BOND CBB  CA                      ! alanine
BOND HB1B CBB      HB2B CBB      HB3B CBB      ! alanine
DOUBLE    O  C
IMPR N  -C CA HN  C CA +N O
DONOR HN N
DONOR HG1A OGA
ACCEPTOR OGA
ACCEPTOR O  C

PRES HYB      0.00
DELE ANGLE  CBA  CA      CBB  ! remove angle between side chains

END

```

Interestingly enough, atoms belonging to the backbone are common to the initial and the final states. Only the side chain of the hybrid residue is affected by the alchemical transformation, and, therefore, requires a dual-topological definition. In addition, considering that only the central amino acid of the blocked tripeptide is modified in the course of the free energy calculation, the standard Ala residues should be invoked in the CHARMM script when building the sequence.

```

read sequence card
* title
*
1
ALA STA ALA

```

It should be mentioned that as the potential energy function is scaled when λ varies from 0 to 1, the side chain of the mutant might flip, thereby resulting in the wrong chirality for the final state of the transformation. To circumvent this undesirable effect, the valence angle formed by the β -carbon atoms of the side chains and the common α -carbon atom should be restrained to zero.

3. Setting up the free energy calculations

Now that the topology of the hybrid molecule is defined, we will detail how the free energy calculation proceeds in NAMD. Execution of the CHARMM script will generate the `psf` topology file used in the MD simulation. One should remember, however, that, on account of the dual-topology paradigm, both the initial state and the final states of the alchemical transformation are present simultaneously. It is, therefore, pivotal that the information about the nature of these states be passed to NAMD, indicating which atoms of the hybrid correspond to $\lambda = 0$, and similarly, which correspond to $\lambda = 1$. This information is given by `fepFile`, a file written in the PDB format, wherein a `-1.00` or `1.00` flag characterizes those atoms of the hybrid molecule that, respectively, vanish or appear in the course of the simulation:

ATOM	1	CI	ZERO	1	-1.167	0.224	0.034	1.00	-1.00	ZERO
ATOM	2	HI1	ZERO	1	-2.133	-0.414	0.000	1.00	-1.00	ZERO
ATOM	3	HI2	ZERO	1	-1.260	0.824	0.876	1.00	-1.00	ZERO
ATOM	4	HI3	ZERO	1	-1.258	0.825	-0.874	1.00	-1.00	ZERO
ATOM	5	CM	ZERO	1	0.001	-0.652	-0.002	1.00	0.00	ZERO
ATOM	6	HM1	ZERO	1	0.000	-1.313	-0.890	1.00	0.00	ZERO
ATOM	7	HM2	ZERO	1	0.005	-1.308	0.889	1.00	0.00	ZERO
ATOM	8	HI	ZERO	1	1.234	0.192	0.000	1.00	-1.00	ZERO
ATOM	9	HF	ZERO	1	-1.237	0.190	0.000	1.00	1.00	ZERO
ATOM	10	CF	ZERO	1	1.289	0.150	-0.078	1.00	1.00	ZERO
ATOM	11	HF1	ZERO	1	2.149	-0.425	-0.001	1.00	1.00	ZERO
ATOM	12	HF2	ZERO	1	1.256	0.837	-0.893	1.00	1.00	ZERO
ATOM	13	HF3	ZERO	1	1.131	0.871	0.940	1.00	1.00	ZERO

The flag that distinguishes between “growing” and “shrinking” atoms can be declared in either the X, Y, Z, O or B column of `fepFile`. In the case of the *zero-sum* ethane \rightarrow ethane transformation, atoms CI, HI1, HI2, HI3 and HI of the initial state vanish, as atoms CF, HF1, HF2, HF3 and HF of the final state appear. A `0.00` flag is assigned to those atoms that are left unchanged as λ varies from 0 to 1.

```
# FEP
```

```
fep                                on
```

```

fepFile          zero.fep
fepCol           B
fepOutFile       zero.fepout
fepOutFreq       5
FepEquilbSteps   3200

```

```
# LOOP OVER LAMBDA-STATES -- TCL
```

```

set step          0.00
set dstep         0.025
dlambda           $dstep

```

```

while {$step <= 1.00} {
    firsttimestep 0
    lambda $step
    lambda2 [expr $step+$dstep]
    run 6400
    set step [expr $step+$dstep]
}

```

TCL scripts allow to set up the protocol of the free energy calculation in a straightforward, user-friendly fashion. In the above example, the potential energy function of the system is scaled from $\lambda = 0$ to $\lambda = 1$ by increments $\delta\lambda = 0.025$, *i.e.* 40 intermediate λ -states or “windows”.⁷ In each “window”, the system is equilibrated over FepEquilbSteps MD steps, *viz.* here 3,200 MD steps, prior to 6,400 MD steps of data collection, from which the ensemble average is evaluated. Using this protocol, the free energy varies smoothly as λ progresses from 0 to 1, and the net free energy change is +0.03 kcal/mol, altogether suggestive that the convergence of the simulation is appropriate.

In the second example, wherein the terminally blocked Ala-Ser-Ala tripeptide is mutated into (Ala)₃, atoms CBA, HB1A, HB2A, OGA and HG1A vanish, as atoms CBB, HB1B, HB2B and HB3B appear:

ATOM	13	N	STA	2	-1.035	-1.123	0.248	1.00	0.00	AHA
ATOM	14	HN	STA	2	-1.281	-1.562	-0.608	1.00	0.00	AHA
ATOM	15	CA	STA	2	0.229	-1.546	0.803	1.00	0.00	AHA
ATOM	16	HA	STA	2	-0.005	-1.844	1.799	1.00	0.00	AHA

ATOM	17	CBA	STA	2	0.933	-2.644	0.012	1.00	-1.00	AHA
ATOM	18	HB1A	STA	2	1.827	-2.977	0.506	1.00	-1.00	AHA
ATOM	19	HB2A	STA	2	0.225	-3.507	-0.122	1.00	-1.00	AHA
ATOM	20	OGA	STA	2	1.263	-2.022	-1.274	1.00	-1.00	AHA
ATOM	21	HG1A	STA	2	1.853	-1.309	-1.234	1.00	-1.00	AHA
ATOM	22	CBB	STA	2	0.937	-2.495	-0.279	1.00	1.00	AHA
ATOM	23	HB1B	STA	2	1.814	-2.963	0.480	1.00	1.00	AHA
ATOM	24	HB2B	STA	2	0.236	-3.490	-0.132	1.00	1.00	AHA
ATOM	25	HB3B	STA	2	1.175	-2.169	-1.095	1.00	1.00	AHA
ATOM	26	C	STA	2	1.207	-0.431	1.069	1.00	0.00	AHA
ATOM	27	O	STA	2	1.752	-0.299	2.140	1.00	0.00	AHA

Because the backbone atoms of the hybrid amino acid are left unchanged throughout the alchemical transformation, a 0.00 flag is assigned to them in the B column of `fepFile`. It is crucial, here, that the declarations in `fepCol` coincides with the list of excluded atoms defined in the CHARMM set-up script. In this particular example, the initial state corresponds to Ala-Ser-Ala, in which the atoms of the L-serine side chain, *viz.* $\lambda = 0$, do not see those of L-alanine, *viz.* $\lambda = 1$. Accordingly, atoms CBA, HB1A, HB2A, OGA and HG1A in `fepFile` should be assigned a -1.00 flag, while the flag for atoms CBB, HB1B, HB2B and HB3B should be 1.00.

```
# FEP
```

```
fep                                on
fepFile                            serine.fep
fepCol                             B
fepOutFile                         serine.fepout
fepOutFreq                         5
FepEquilbSteps                     6400
```

```
# LOOP OVER LAMBDA-STATES -- TCL
```

```
set step                          0.00
set dstep                         0.025
dlambda                           $dstep
```

```
while {$step <= 1.00} {
    firsttimestep 0
    lambda $step
```

```

lambda2 [expr $step+$dstep]
run 12800
set step [expr $step+$dstep]
}

```

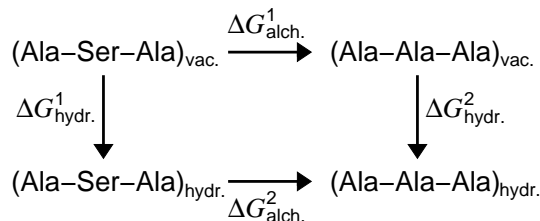


Figure 3: Thermodynamic cycle used in the $\text{Ala-Ser-Ala} \rightarrow (\text{Ala})_3$ alchemical transformation. The vertical arrows correspond to the hydration of the wild-type tripeptide and its mutant. The horizontal arrows correspond to the point mutation in bulk water and *in vacuo*, so that: $\Delta G_{\text{alch.}}^2 - \Delta G_{\text{alch.}}^1 = \Delta G_{\text{hydr.}}^2 - \Delta G_{\text{hydr.}}^1$.

Just like in the *zero-sum* ethane \rightarrow ethane alchemical transformation, use is made here of 40 “windows” to connect the initial state, *viz.* Ser, to the final state, *viz.* Ala, of the mutation. Here, each individual λ -state consists, however, of 6,400 MD steps of equilibration, followed by 12,800 MD steps of data collection, from which the ensemble average is computed. In order to estimate the free energy change involved in the point mutation of the hydrated tripeptide, the same simulation should be carried out in bulk water and *in vacuo*, to close the thermodynamic cycle of Figure 3.⁸ Adopting the same sampling protocol for the two legs of the mutation, the free energy difference for the hydrated state is +1.80 kcal/mol, and −5.57 kcal/mol for the isolated state, thus yielding a net free energy change equal to +7.37 kcal/mol for the overall $\text{Ala-Ser-Ala} \rightarrow (\text{Ala})_3$ transformation. In sharp contrast with the preceding example, the accuracy of the free energy difference associated to this alchemical transformation inherently depends upon the quality of the potential energy function utilized. Deconvoluting the error in such calculations in terms of (i) inadequacy of the force field, and, (ii) sampling of inappropriate length, constitutes a daunting task. A close agreement with the experimental value may very well be the fortuitous result of an insufficient sampling and a poorly parameterized potential energy function.^{9,2} Here, the smoothly changing free energy and the nice accord with the experimental estimate of +7.02 kcal/mol,^{10,11} not only suggest that the simulation has converged, but also that the parameters used are well-adapted to the problem tackled.

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